

Rayat Shikshan Sanstha's  
**Karmaveer Bhaurao Patil Mahavidyalaya Pandharpur**  
(Autonomous)  
**Department of Chemistry**  
**B. Sc. I, (2020-21)**  
**Report on**  
**Slow and Advanced Learners Scheme**

The Scheme of Slow and Advanced Learners has been implemented in the year 2020-21 for B.Sc. I year Students.

In this scheme the selection of students as Slow Learners and Advanced Learners is done on the basis of the marks secured by the students in the first unit test conducted on 08/09/2021. The unit test was for 10 marks. Those who obtained less or equal to 4 marks have been selected as slow learner and those who secured more than or equal to 7 marks have been selected as advanced learners.

10 students have been selected as advanced learners. For these advanced learners special attention has been paid and activities such as extra lectures, group discussions and problem solving session have been conducted to further boost their performance.

11 students have been selected as slow learners. For these slow learners special attention has been paid in order to improve their score and understanding of the subject. Activities such as special counselling, unit tests and sessions on discussion of how to solve questions in examination were conducted for slow learners.

All these activities have been conducted online due to COVID-19 pandemic situation.

  
**Head**  
Department of Chemistry  
K.B.P. Mahavidyalaya, Pandharpur

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**Department of Chemistry**

**Slow and Advanced Learners**

**Notice**

**Date 04 /09/ 2020**

All the student of B.Sc.-I( Chemistry) are selected as Slow and Advanced Learners are here informed that they should attend the activities which are groining to be conducted by the Department of Chemistry during academic year 2020-2021.

  
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**Department of Chemistry**

**B. Sc. -I**

**Slow and Advanced Learners (2020-21)**

<b>Sr. No.</b>	<b>Full name of Student</b>	<b>Marks</b>
1.	Ganesh Balu Sutar	1 / 10
2.	Sapate Aniruddha Uttam	3 / 10
3.	Sayali Navanath Mane	3 / 10
4.	Shelake Rajkumar	3 / 10
5.	Vishnavi Sanjay Bidvai	3 / 10
6.	Hajare Tejas Prakash	4 / 10
7.	Kale Priti	4 / 10
8.	Shubham Bharat Kshirsagar	4 / 10
9.	Ghodake Samarth Nanasaheb	4 / 10
10.	Abhijeet Gajare	4 / 10
11.	Monika Shivaji Pise	4 / 10
12.	Saraswati Nagesh Korake	4 / 10
13.	Pranali Mohan Korake	5 / 10
14.	Onkar Kalagate	5 / 10
15.	Neha Sanjay Yelmar	5 / 10
16.	Meghraj Deshmukh	5 / 10
17.	Dandade Snehal Satyawan	5 / 10
18.	Suraj Satish Bhosale	5 / 10
19.	Waydande Reshma Kisan	5 / 10
20.	Sakshi Charudatta Patil	5 / 10
21.	Korake Amruta Mahadev	5 / 10
22.	Pooja Dinesh Hodge	5 / 10
23.	Pratidnya Gutal.	5 / 10
24.	Chakuli Chavare	5 / 10
25.	Rohan Aba Ghutukade	6 / 10

26.	Harshraj Kalunge.	6 / 10
27.	Shubham Haridas Gaikwad	6 / 10
28.	Gaikwad Rajlakshmi Dhanaji	6 / 10
29.	Sujit Nandkumar Sartape	6 / 10
30.	Aishwarya Savata Katkar.	6 / 10
31.	Surekha Jagannath Gorad	6 / 10
32.	Nishigandha Somnath Kshirsagar	6 / 10
33.	Sayli Bhimrao Bansode	6 / 10
34.	Jadhav Yashodip Raghunandan	6 / 10
35.	Pooja Sunil Korake	6 / 10
36.	Rutuja Ramesh Barale	6 / 10
37.	Rutuja Maruti Kedar	6 / 10
38.	Dandade Yuvraj Santosh	6 / 10
39.	Vaishnavi Kulkarni	7 / 10
40.	Vaishnavi Balasaheb Khule	7 / 10
41.	Kishori Kashinath Kamble	7 / 10
42.	Smita Dattatray Magade	7 / 10
43.	Chavan Vaishnavi Vishnu	7 / 10
44.	Shubham Angad Patole	7 / 10
45.	Pravina Anna Yadav	7 / 10
46.	Rutuja Sahadev Koli	8 / 10
47.	Prathamesh Sanjay Yadav	8 / 10
48.	Tanvirbanu Yasin Shaikh	8 / 10
49.	Gaikwad Gitanjali Atmaram	9 / 10
50.	Purva Deepak Survase	9 / 10
51.	Kajal Hari Ghadge	9 / 10
52.	Nagesh Gavali	9 / 10
53.	Vidya Dasharath Phalake	10 / 10
54.	Nawale Sonali Dilip	10 / 10

  
Head

Department of Chemistry  
K.B.P. Mahavidyalaya, Pandharpur





**Rayat Shikshan Sanstha's**  
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**Department of Chemistry**  
**Mark List**  
**Advanced Learners**  
**B.Sc.I**

**Date:- 09/09/2020**

**Total Marks-10**

<b>Sr. No.</b>	<b>Full name of Student</b>	<b>Marks</b>
1.	Pravina Anna Yadav	7 / 10
2.	Rutuja Sahadev Koli	8 / 10
3.	Prathamesh Sanjay Yadav	8 / 10
4.	Tanvirbanu Yasin Shaikh	8 / 10
5.	Gaikwad Gitanjali Atmaram	9 / 10
6.	Purva Deepak Survase	9 / 10
7.	Kajal Hari Ghadge	9 / 10
8.	Nagesh Gavali	9 / 10
9.	Vidya Dasharath Phalake	10 / 10
10.	Nawale Sonali Dilip	10 / 10

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**B.Sc.I**

**Attendance**  
**Advanced Learners**

w.e.f. September 09/09/2020

From 10/09/2020 to 21/09/2020

Sr.No	Full Name of Student	10/09	11/09	14/09	15/09	16/09	17/09	18/09	21/09
1.	Pravina Anna Yadav	P	P	P	P	P	P	P	P
2.	Rutuja Sahadev Koli	P	P	P	P	P	P	P	P
3.	Prathamesh Sanjay Yadav	P	P	A	A	P	P	P	P
4.	Tanvirbanu Yasin Shaikh	P	P	P	P	P	P	P	P
5.	Gaikwad Gitanjali Atmaram	P	P	P	P	P	P	P	P
6.	Purva Deepak Survase	P	P	P	A	P	P	P	P
7.	Kajal Hari Ghadge	P	P	P	P	P	P	P	P
8.	Nagesh Gavali	P	P	P	P	P	P	P	P
9.	Vidya Dasharath Phalake	P	P	P	P	A	P	P	P
10.	Nawale Sonali Dilip	P	P	P	P	P	P	P	P

  
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**Test for Slow learners**

**B.Sc.I**

**Date:- 08/09/2021**

**Marks- 10**

**Q1) Select and write the answer of the following question**

**Marks -10**

- 
- 1) Which charge present on electron?  
a) Positive    b) Negative    c) Neutral    d) All of these
  - 2) Which fundamental particles present in nucleus  
a) Electron and Protons                      b) Protons and Neutrons  
c) Only protons                                  d) Only neutrons
  - 3) Atomic number of carbon.....is  
a) 5              b) 3              c) 4              d) 6
  - 4) Nucleophiles are.....  
a) Electron rich species                      b) Electron deficient species  
c) Electron sufficient species              d) All of these
  - 5) How many blocks in periodic table?  
a) 2              b) 5              c) 4              d) 6
  - 6) What is the valiancy of Carbon ?  
a) 1              b) 2              c) 3              d) 4
  - 7) According VSEPR theory what is the shape of Water molecule?  
a) V-shaped              b) W- shaped              c) X- shaped              d) Z- shaped
  - 8) What is hybridization of carbon in benzene.?  
a) SP              b)  $sp^2$               c)  $SP^3$               d)  $SP^3d$
  - 9) One bond is equal to how many electrons ?  
a) One              b) Two              c) Three              d) Four
  - 10) What is general molecular formula of alkynes?  
a)  $C_nH_{2n}$               b)  $C_nH_{2n+2}$               c)  $C_nH_{2n-2}$               d)  $C_nH_{2n+1}$



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**Mark List**  
**B.Sc.I**  
**Slow Learner**

**Date:- 09/09/2020**

**Total Marks-10**

<b>Sr. No.</b>	<b>Full name of Student</b>	<b>Score</b>
1.	Ganesh Balu Sutar	1 / 10
2.	Sapate Aniruddha Uttam	3 / 10
3.	Sayali Navanath Mane	3 / 10
4.	Shelake Rajkumar	3 / 10
5.	Vishnavi Sanjay Bidvai	3 / 10
6.	Hajare Tejas Prakash	4 / 10
7.	Kale Priti	4 / 10
8.	Shubham Bharat Kshirsagar	4 / 10
9.	Ghodake Samarth Nanasaheb	4 / 10
10.	Abhijeet Gajare	4 / 10
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**B.Sc.I**  
**Attendance**  
**Slow Learner Learners**

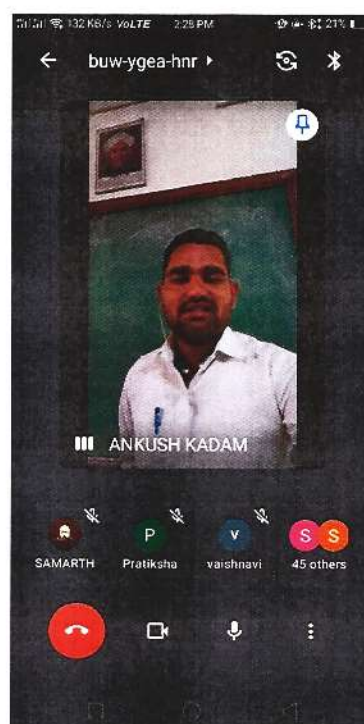
w.e.f. September 09/09/2020

From 10/09/2020 to 21/09/2020

Sr. No.	Full name of Student	10/09	11/09	14/09	15/09	16/09	17/9	18/09	21/09
1.	Ganesh Balu Sutar	P	P	P	P	P	P	P	P
2.	Sapate Aniruddha Uttam	P	P	P	P	P	P	P	P
3.	Sayali Navanath Mane	P	A	P	P	P	P	P	P
4.	Shelake Rajkumar	P	P	P	P	P	P	P	P
5.	Vishnavi Sanjay Bidvai	P	P	P	P	P	P	P	P
6.	Hajare Tejas Prakash	P	P	P	P	P	P	P	P
7.	Kale Priti	P	P	P	P	P	P	P	P
8.	Shubham Bharat Kshirsagar	P	P	A	A	P	P	P	P
9.	Ghodake Samarth Nanasaheb	P	P	P	P	P	P	P	P
10.	Abhijeet Gajare	P	P	P	P	A	P	P	P
11.	Monika Shivaji Pise	P	P	P	P	P	P	A	P

  
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Department of Chemistry  
B.Sc.I  
Advanced and Slow Learner  
Lecture conducted through online mode



  
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